

Observational Indications of Two Primary Processes Producing Elements from Sr to Eu

INT workshop

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ARI ITA LSW

Outline - From Stellar spectra to the r-process(es?)

- Telescopes and stellar spectra
- Stellar abundances and uncertainties
- Observational indications of a 2nd process
- Meteorites and presolar grains
- Disentangling the primary processes

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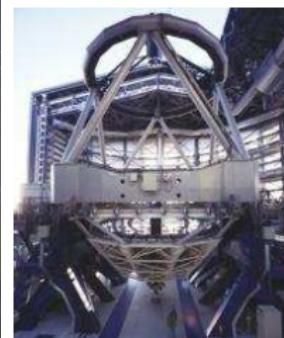
VLT/UVES and LAMOST

Very Large Telescope (VLT) - 8-m mirror

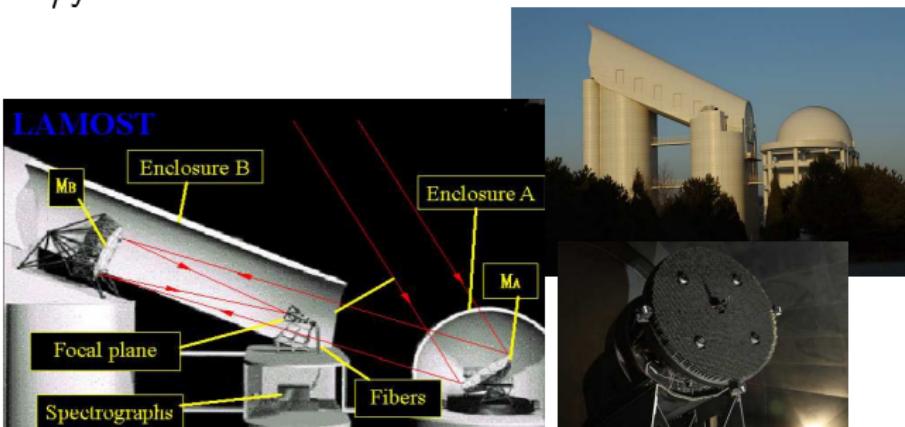


Fig. 1—The essential components of an astronomical spectrograph

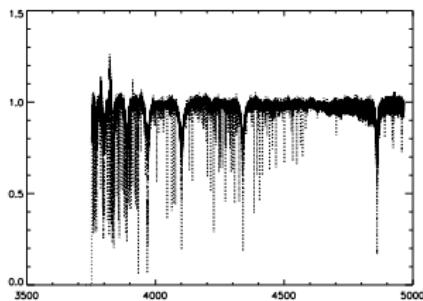
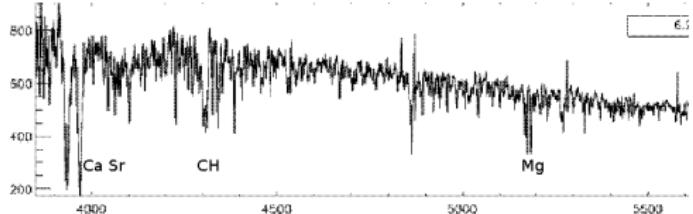
Simple sketch of a spectrograph — Massey et al.



Large Sky Area Multi-Object Fiber Spectroscopic Telescope (LAMOST) — 4-m mirror, 4000 fibres → 10000 stars/night or $2 \cdot 10^6$ stars/year



LAMOST vs UVES spectra



LAMOST (low resolution $R \sim 1800$) and ESO VLT (UVES - high resolution $R \sim 40000$)

Important: Sr may be the only heavy element for which we will be able to derive abundances in low-resolution spectra.

Telescopes
ooo

Abundances
●ooooooo

Applications
ooo

Heavy elements
ooo

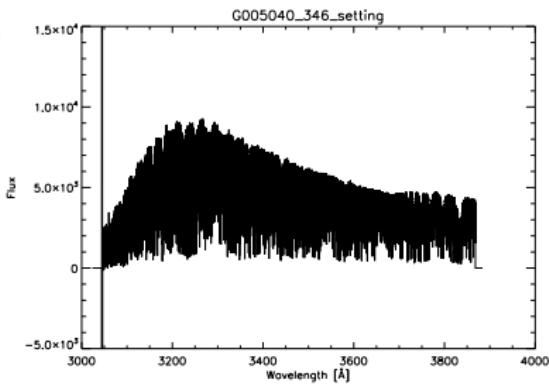
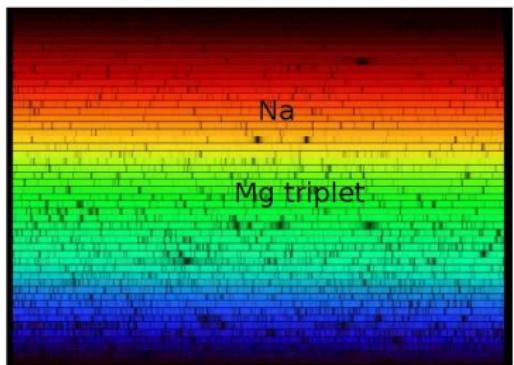
2. r-process
oooooooooooo

Yields
ooo

Meteorites
ooooooo

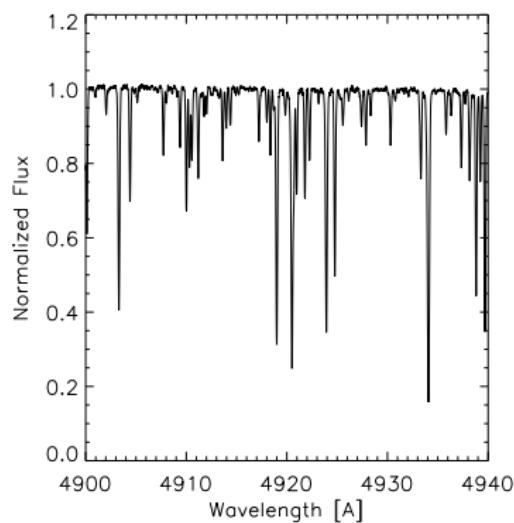
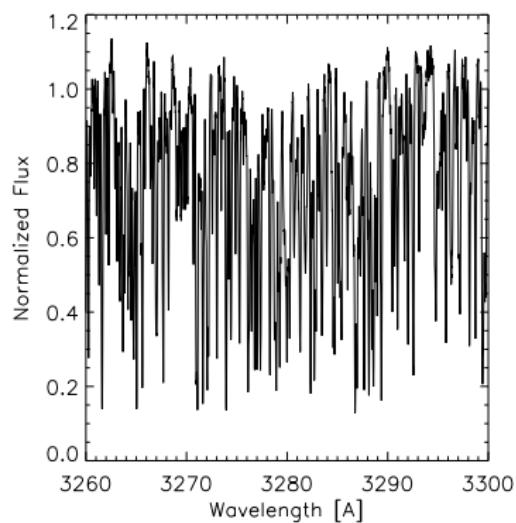
Separating processes
oooooooooooo

Stellar spectra – 2D to 1D



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Visual versus near-UV spectral range

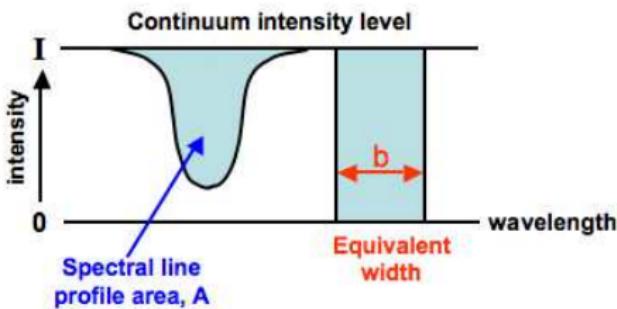
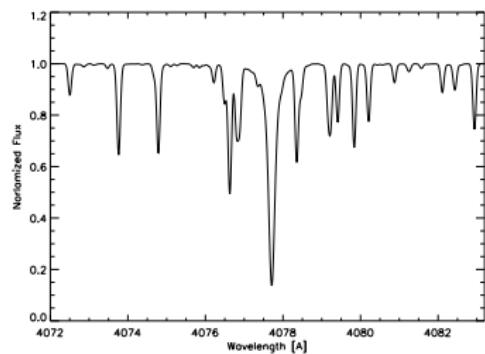


A metal-poor giant star (HD122956): Temperature/gravity/[Fe/H] =
4700K/1.5/-1.45



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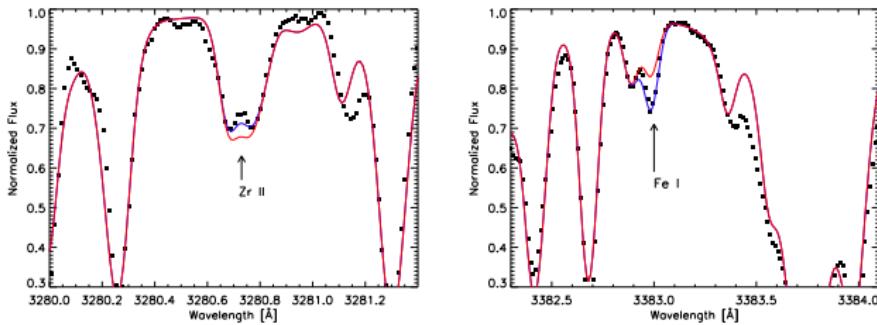
Stellar spectra and equivalent width (W)



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The importance of atomic data; Abundance - log gf relation

$$\log W = \log(\text{const}) + \log(A) + \log(gf\lambda) - \theta\chi - \log(\kappa) \quad (1)$$

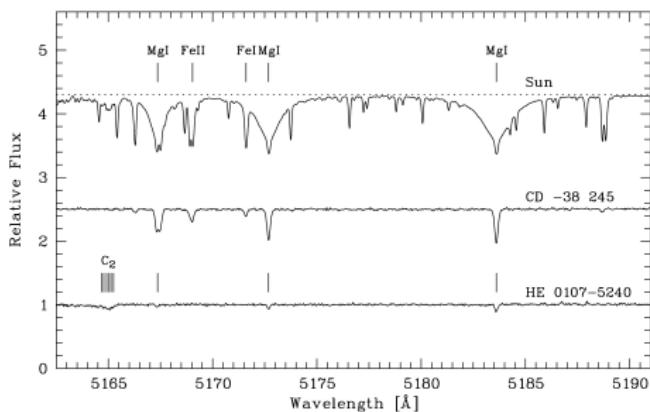


Hansen et al., 2012

Since the UV-region of the spectra is crowded we have to carry out spectral synthesis on line lists with accurate atomic data.

Stellar spectra, abundances, and $[\text{Fe}/\text{H}]$

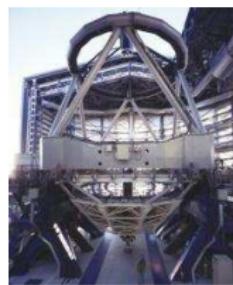
$$[\text{Fe}/\text{H}] \equiv \log(N_{\text{Fe}}/N_{\text{H}})_* - \log(N_{\text{Fe}}/N_{\text{H}})_{\odot} \quad (2)$$



Top: Solar ($[\text{Fe}/\text{H}] = 0$) spectrum – Mg triplet. Bottom: Star with $[\text{Fe}/\text{H}] \sim -5$. Christlieb +2004

Some of the most metal-poor stars!
See the next talk by Terese Hansen

Observable elements - with high-resolution instruments



Periodic Table of the Elements

 www.elementsdatabase.com

Blue: ground based observations, green: space, yellow: isotopic abundances



Record holding star
 - CS31082-001
 Abundances
 of almost 70 elements,
 37 of which are heavy elements.

Siqueira Mello et al. 2013

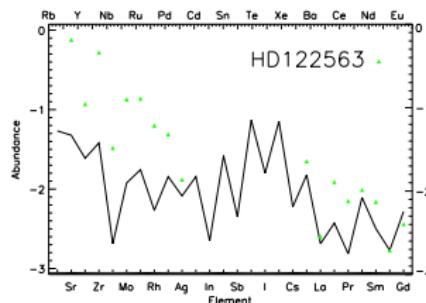
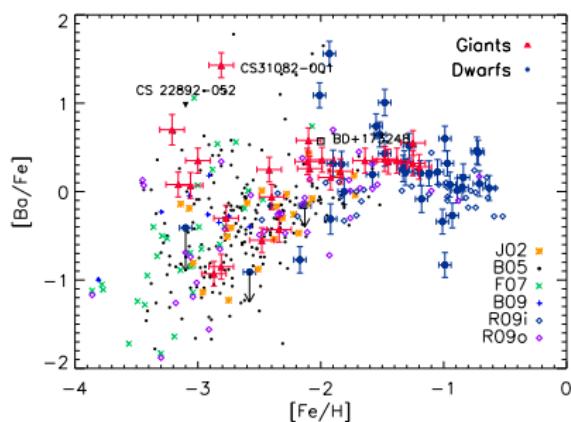
Table 1. LTE abundances in CS 31082-001 as derived from previous works, from the present paper, and our adopted final abundances.

El.	Z	A(X) (1)	A(X) (2)	A(X) (3)	A(X) This Work	A(X) adopted	[X/Fe] adopted
Ge	32	—	—	—	+0.10	+0.10±0.21	-0.55
Sr	38	+0.72	—	—	—	+0.72±0.10	0.73
Y	39	-0.23	—	—	-0.15	-0.19±0.07	0.53
Zr	40	+0.43	—	—	+0.55	+0.49±0.08	0.84
Nb	41	-0.55	—	—	-0.52	-0.54±0.12	0.97
Mo	42	—	—	—	-0.11	-0.11±0.13	0.90
Ru	44	+0.36	—	—	+0.36	+0.36±0.12	1.45
Rh	45	-0.42	—	—	-0.42	-0.42±0.12	1.39
Pd	46	-0.05	—	—	-0.09	-0.09±0.07	1.18
Ag	47	-0.81	—	—	-0.84	-0.84±0.21	1.15
Ba	56	+0.40	—	—	—	+0.40±0.14	1.16
La	57	-0.60	-0.62	—	—	-0.62±0.05	1.17
Ce	58	-0.31	-0.29	—	-0.31	-0.29±0.05	1.03
Pr	59	-0.86	-0.79	—	—	-0.79±0.05	1.38
Nd	60	-0.13	-0.15	—	-0.21	-0.15±0.05	1.33
Sm	62	-0.51	-0.42	—	-0.42	-0.42±0.05	1.51
Eu	63	-0.76	-0.72	—	-0.75	-0.72±0.05	1.69
Gd	64	-0.27	-0.21	—	-0.29	-0.21±0.05	1.61
Tb	65	-1.26	-1.01	—	-1.00	-1.01±0.05	1.64
Dy	66	-0.21	-0.07	—	-0.12	-0.07±0.05	1.73
Ho	67	—	-0.80	—	—	-0.80±0.06	1.62
Er	68	-0.27	-0.30	—	-0.31	-0.30±0.05	1.67
Tm	69	-1.24	-1.15	—	-1.18	-1.15±0.05	1.64
Yb	70	—	-0.41	—	—	-0.41±0.11	1.66
Lu	71	—	—	—	-1.08	-1.08±0.13	1.73
Hf	72	-0.59	-0.72	—	-0.73	-0.72±0.05	1.33
Ta	73	—	—	—	-1.60	-1.60±0.23	1.47
W	74	—	—	—	-0.90	-0.90±0.24	0.92
Re	75	—	—	—	-0.21	-0.21±0.21	2.45
Os	76	+0.43	—	+0.18	—	+0.18±0.07	1.72
Ir	77	+0.20	—	+0.20	—	+0.20±0.07	1.72
Pt	78	—	—	+0.30	—	+0.30±0.23	1.46
Au	79	—	—	-1.00	—	-1.00±0.34	0.89
Pb	82	—	—	-0.65	—	-0.65±0.19	0.25
Bi	83	—	—	-0.40	—	-0.40±0.33	1.83
Th	90	-0.98	—	—	—	-0.98±0.13	1.84
U	92	-1.92	—	—	—	-1.92±0.17	1.68

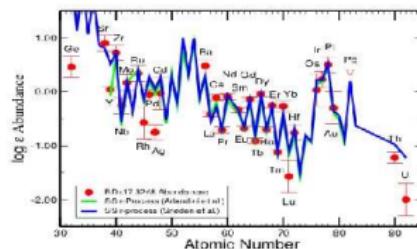
References. (1) Hill et al. (2002), (2) Sneden et al. (2009), (3) Barbuy et al. (2011).

What can we learn from stellar abundances?

- HD122563 - proto LEPP star
 - Large star-to-star scatter for n-capture elements (e.g. Sr and Ba)



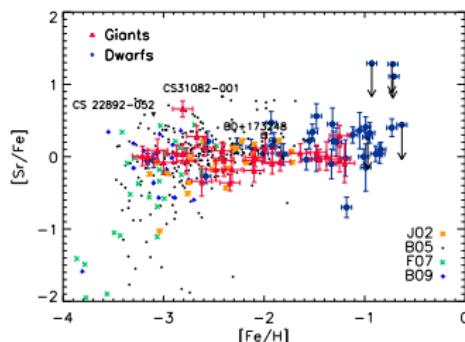
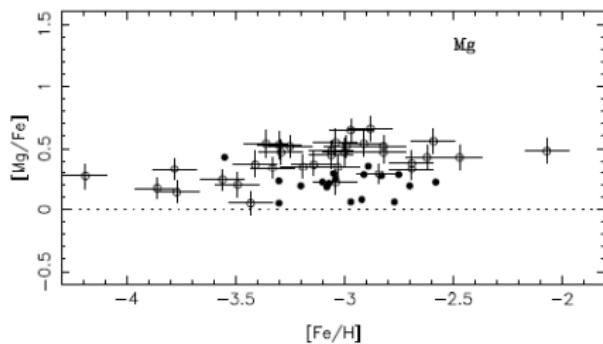
Hansen 2011 & Cowan +2011 (below)



Abundance star-to-star scatter and the 2nd r-process

- α - elements show a very low scatter
 - Sr shows a very large scatter

Bonifacio et al. 2009



Hansen et al., 2012

Selected elements

Periodic Table of Elements

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18																																																																				
1 H Atomic # 1 Symbol H Name Hydrogen Atomic Mass 1.008	2 He Atomic # 2 Symbol He Name Helium Atomic Mass 4.003	3 Li Atomic # 3 Symbol Li Name Lithium Atomic Mass 6.941	4 B Atomic # 4 Symbol B Name Boron Atomic Mass 10.81	5 N Atomic # 5 Symbol N Name Nitrogen Atomic Mass 14.01	6 O Atomic # 6 Symbol O Name Oxygen Atomic Mass 15.999	7 F Atomic # 7 Symbol F Name Fluorine Atomic Mass 18.998	8 Ne Atomic # 8 Symbol Ne Name Neon Atomic Mass 20.18	9 Na Atomic # 9 Symbol Na Name Sodium Atomic Mass 22.99	10 Mg Atomic # 12 Symbol Mg Name Magnesium Atomic Mass 24.31	11 Al Atomic # 13 Symbol Al Name Aluminum Atomic Mass 26.98	12 Si Atomic # 14 Symbol Si Name Silicon Atomic Mass 28.09	13 P Atomic # 15 Symbol P Name Phosphorus Atomic Mass 30.97	14 S Atomic # 16 Symbol S Name Sulfur Atomic Mass 32.07	15 Cl Atomic # 17 Symbol Cl Name Chlorine Atomic Mass 35.45	16 Ar Atomic # 18 Symbol Ar Name Argon Atomic Mass 39.95	17 K Atomic # 19 Symbol K Name Potassium Atomic Mass 39.09	18 Ca Atomic # 20 Symbol Ca Name Calcium Atomic Mass 40.08	19 Sc Atomic # 21 Symbol Sc Name Scandium Atomic Mass 44.96	20 Ti Atomic # 22 Symbol Ti Name Titanium Atomic Mass 47.87	21 V Atomic # 23 Symbol V Name Vanadium Atomic Mass 50.94	22 Cr Atomic # 24 Symbol Cr Name Chromium Atomic Mass 51.99	23 Mn Atomic # 25 Symbol Mn Name Manganese Atomic Mass 54.94	24 Fe Atomic # 26 Symbol Fe Name Iron Atomic Mass 55.85	25 Co Atomic # 27 Symbol Co Name Cobalt Atomic Mass 58.93	26 Ni Atomic # 28 Symbol Ni Name Nickel Atomic Mass 58.7	27 Cu Atomic # 29 Symbol Cu Name Copper Atomic Mass 63.55	28 Zn Atomic # 30 Symbol Zn Name Zinc Atomic Mass 65.4	29 Ga Atomic # 31 Symbol Ga Name Gallium Atomic Mass 69.72	30 Ge Atomic # 32 Symbol Ge Name Germanium Atomic Mass 72.63	31 As Atomic # 33 Symbol As Name Arsenic Atomic Mass 75.0	32 Se Atomic # 34 Symbol Se Name Selenium Atomic Mass 78.96	33 Br Atomic # 35 Symbol Br Name Bromine Atomic Mass 79.90	34 Kr Atomic # 36 Symbol Kr Name Krypton Atomic Mass 83.81	35 Xe Atomic # 37 Symbol Xe Name Xenon Atomic Mass 131.3	36 Rb Atomic # 38 Symbol Rb Name Rubidium Atomic Mass 85.47	37 Sr Atomic # 39 Symbol Sr Name Strontium Atomic Mass 87.62	38 Y Atomic # 40 Symbol Y Name Yttrium Atomic Mass 88.91	39 Zr Atomic # 41 Symbol Zr Name Zirconium Atomic Mass 91.22	40 Nb Atomic # 42 Symbol Nb Name Niobium Atomic Mass 92.91	41 Mo Atomic # 43 Symbol Mo Name Molybdenum Atomic Mass 95.94	42 Tc Atomic # 44 Symbol Tc Name Technetium Atomic Mass 97.92	43 Ru Atomic # 45 Symbol Ru Name Ruthenium Atomic Mass 101.07	44 Rh Atomic # 46 Symbol Rh Name Rhodium Atomic Mass 102.91	45 Pd Atomic # 47 Symbol Pd Name Palladium Atomic Mass 106.42	46 Ag Atomic # 48 Symbol Ag Name Silver Atomic Mass 107.87	47 Cd Atomic # 49 Symbol Cd Name Cadmium Atomic Mass 112.41	48 In Atomic # 50 Symbol In Name Indium Atomic Mass 114.82	49 Sn Atomic # 51 Symbol Sn Name Tin Atomic Mass 118.71	50 Sb Atomic # 52 Symbol Sb Name Antimony Atomic Mass 121.76	51 Te Atomic # 53 Symbol Te Name Tellurium Atomic Mass 127.60	52 I Atomic # 54 Symbol I Name Iodine Atomic Mass 126.90	53 Xe Atomic # 55 Symbol Xe Name Xenon Atomic Mass 131.3	54 Cs Atomic # 56 Symbol Cs Name Cesium Atomic Mass 132.91	55 Ba Atomic # 57 Symbol Ba Name Barium Atomic Mass 137.34	56 Hf Atomic # 72 Symbol Hf Name Hafnium Atomic Mass 178.49	57-71 Ta Atomic # 73 Symbol Ta Name Tantalum Atomic Mass 182.95	58 W Atomic # 74 Symbol W Name Tungsten Atomic Mass 183.84	59 Re Atomic # 75 Symbol Re Name Rhenium Atomic Mass 186.21	60 Os Atomic # 76 Symbol Os Name Osmium Atomic Mass 190.23	61 Ir Atomic # 77 Symbol Ir Name Iridium Atomic Mass 192.22	62 Pt Atomic # 78 Symbol Pt Name Platinum Atomic Mass 195.08	63 Au Atomic # 79 Symbol Au Name Gold Atomic Mass 196.97	64 Hg Atomic # 80 Symbol Hg Name Mercury Atomic Mass 200.59	65 Tl Atomic # 81 Symbol Tl Name Thallium Atomic Mass 204.42	66 Pb Atomic # 82 Symbol Pb Name Lead Atomic Mass 207.2	67 Bi Atomic # 83 Symbol Bi Name Bismuth Atomic Mass 208.98	68 Po Atomic # 84 Symbol Po Name Polonium Atomic Mass 210.0	69 At Atomic # 85 Symbol At Name Astatine Atomic Mass 210.0	70 Rn Atomic # 86 Symbol Rn Name Radon Atomic Mass 222.0	71 Fr Atomic # 87 Symbol Fr Name Francium Atomic Mass 223.0	72-101 Rb Atomic # 88-101 Symbol Rb Name Rubidium Atomic Mass 88.91-101.07	73 Rb Atomic # 88 Symbol Rb Name Rubidium Atomic Mass 88.91	74 Fr Atomic # 89 Symbol Fr Name Francium Atomic Mass 223.0	75 Rb Atomic # 90 Symbol Rb Name Rubidium Atomic Mass 88.91	76 Rb Atomic # 91 Symbol Rb Name Rubidium Atomic Mass 88.91	77 Rb Atomic # 92 Symbol Rb Name Rubidium Atomic Mass 88.91	78 Rb Atomic # 93 Symbol Rb Name Rubidium Atomic Mass 88.91	79 Rb Atomic # 94 Symbol Rb Name Rubidium Atomic Mass 88.91	80 Rb Atomic # 95 Symbol Rb Name Rubidium Atomic Mass 88.91	81 Rb Atomic # 96 Symbol Rb Name Rubidium Atomic Mass 88.91	82 Rb Atomic # 97 Symbol Rb Name Rubidium Atomic Mass 88.91	83 Rb Atomic # 98 Symbol Rb Name Rubidium Atomic Mass 88.91	84 Rb Atomic # 99 Symbol Rb Name Rubidium Atomic Mass 88.91	85 Rb Atomic # 100 Symbol Rb Name Rubidium Atomic Mass 88.91	86 Rb Atomic # 101 Symbol Rb Name Rubidium Atomic Mass 88.91

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

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Sample, Method, and Formation Process:

- Sample consists of 71 stars, 42 dwarfs and 29 giants
- Enhanced as well as 'normal' stars ($-3.3 < [\text{Fe}/\text{H}] < -0.6$)
- UVES and HIRES (high resolution data)
- MARCS 1D atmospheres & MOOG¹ synthetic spectrum code
- Element and formation process:
 - Sr 85% s-process (weak s-process/ α -rich/p-rich)
 - Y 92% s-process (weak s)
 - Zr 83% s-process (less weak s)
 - Mo 50% s-process (the remaining 50% is from r+p-process)
 - Ru 30% s-process (70% weak r-process?)
 - Pd 46% s-process (54% r-process - some 'weak' r?)
 - Ag 79% r-process ('weak' r?)
 - Ba 81% main s-process (AGB stars)
 - Eu 94% main r (Arlandini + 1999)

¹Sneden 73, version 2010, Assuming LTE

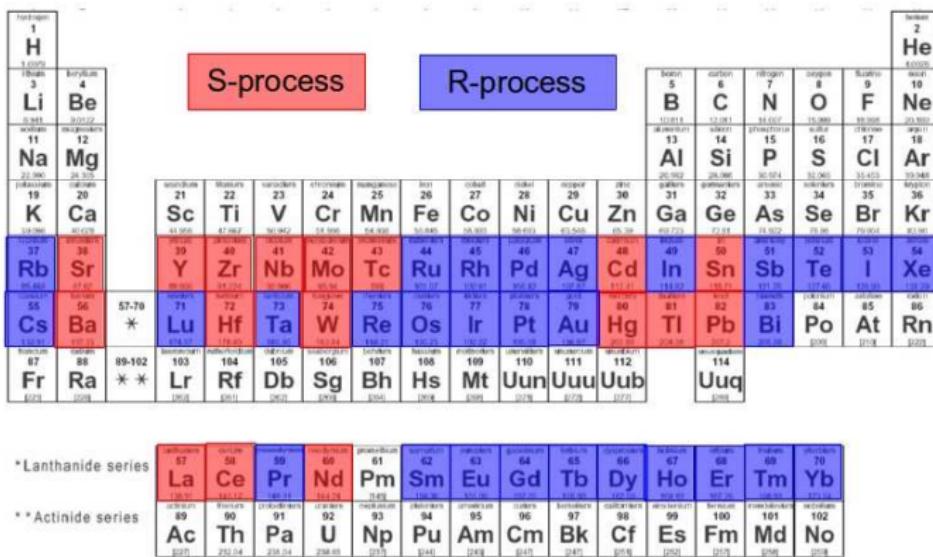
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¹Sneden 73, version 2010, Assuming LTE

Correlations

r- and s-process elements (Arlandini+ 1999)

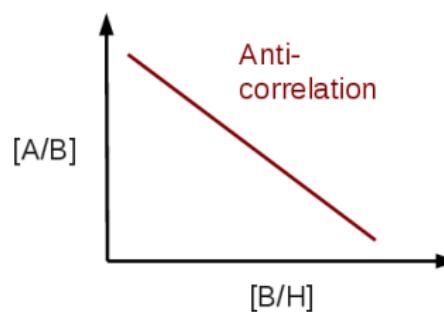
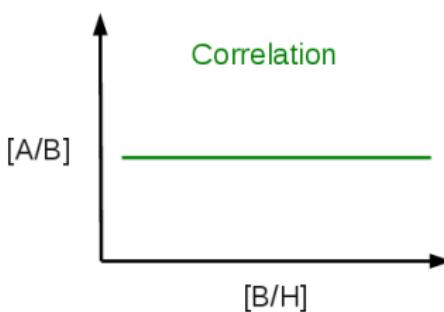


Correlations

Correlation = Anticorrelation

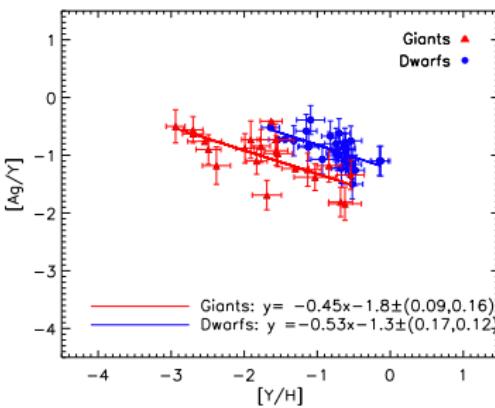
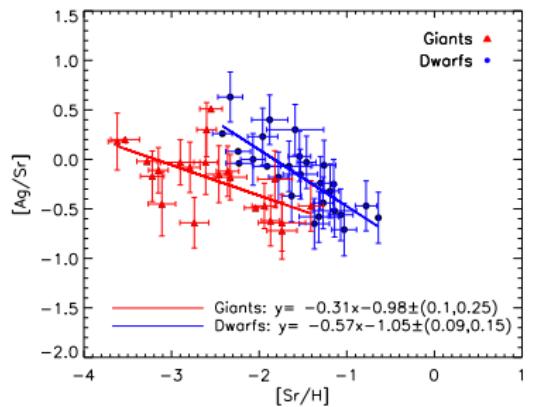
If two elements are created by the same process, they most likely grow in the same way (correlate).

Elements ($38 < Z < 50$) are generally found to anti-correlate with $Z > 56$ elements (Burris et al. 2000, Montes et al. 2007, Francois et al 2007)

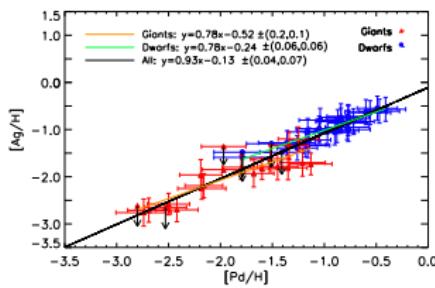
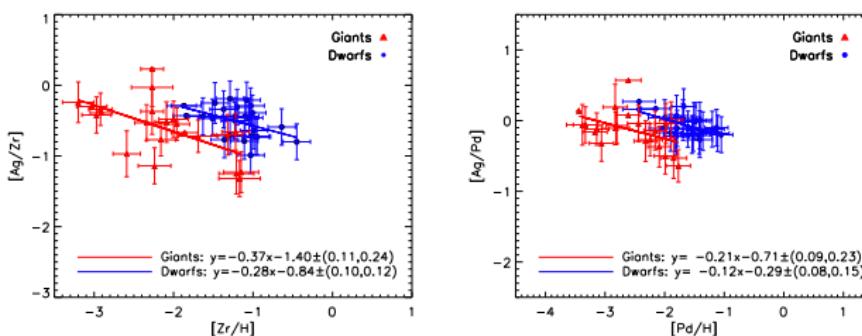


Weak s-process elements - Sr (85%) and Y (92%) Arlandini et al 1999

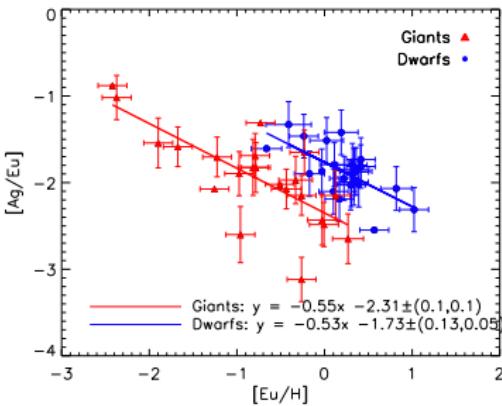
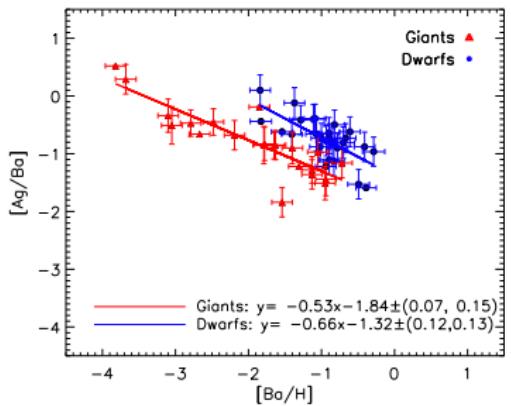
Hansen et al. 2012



Weak s-process and weak r-process/LEPP elements



Main s-process and main r-process elements - Ba (81%) and Eu (94%)

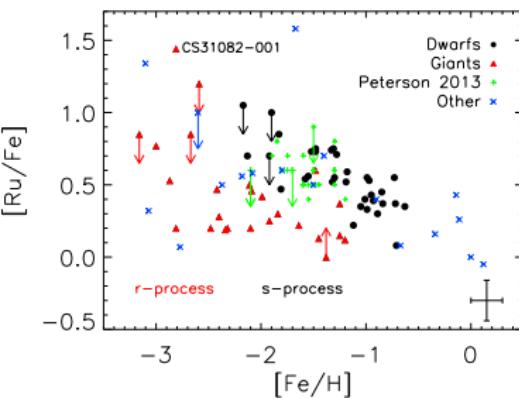
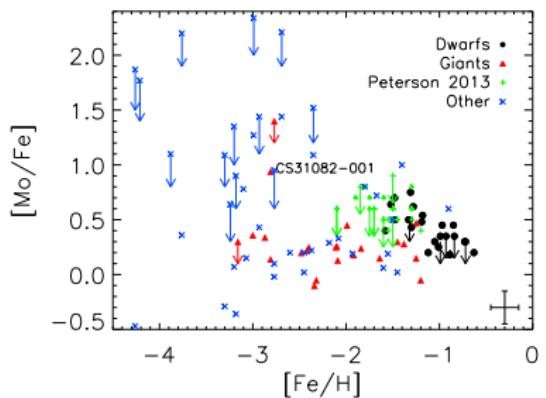


Hansen et al, 2012

This is why silver is interesting:

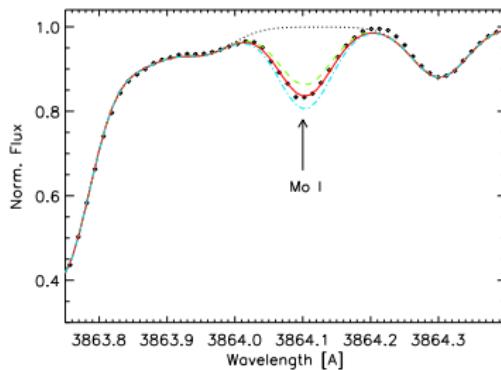
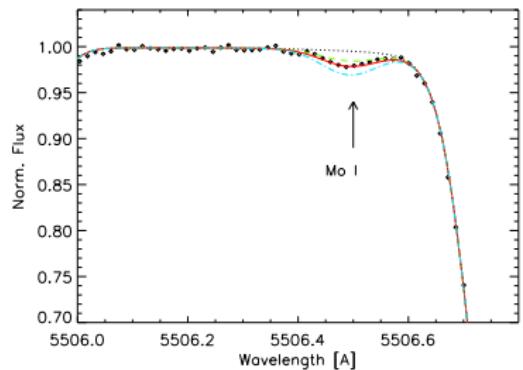
Ag (and Pd) is produced by a second 'weak' r-process/LEPP

Mo and Ru may also be created by this 'LEPP' process



Hansen et al. 2014

The challenge: Deriving abundances from stars that are not enhanced in heavy elements.

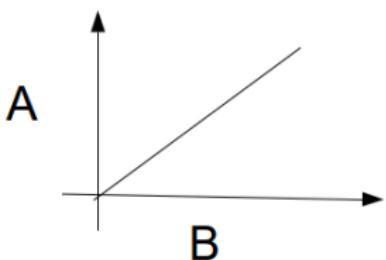


High-quality observations are needed in the near-UV spectral range
- almost impossible with fibre-based instruments.

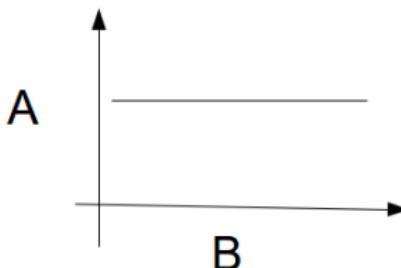
What can we learn about Mo and Ru?

A more direct approach to test if two elements (A, B) correlate

Correlation

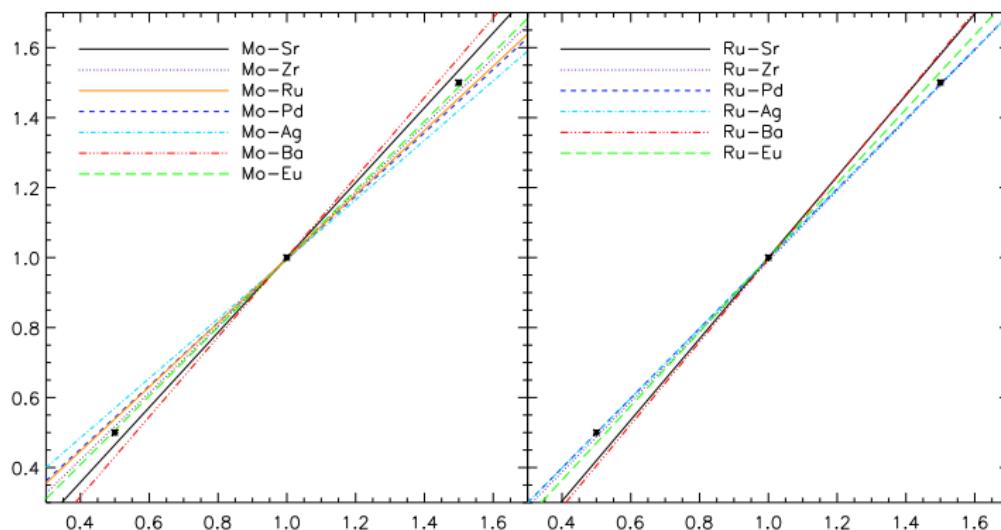


Anti-correlation



What can we learn about Mo and Ru?

Fitting the entire sample = 1 process creates it all..?

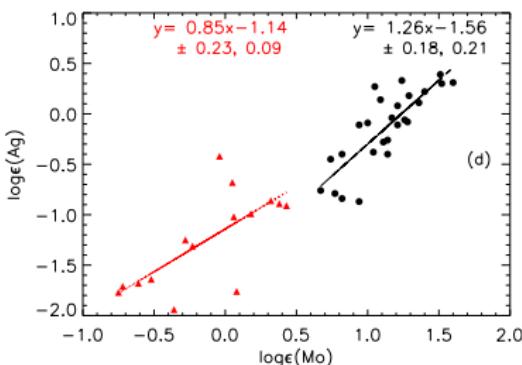
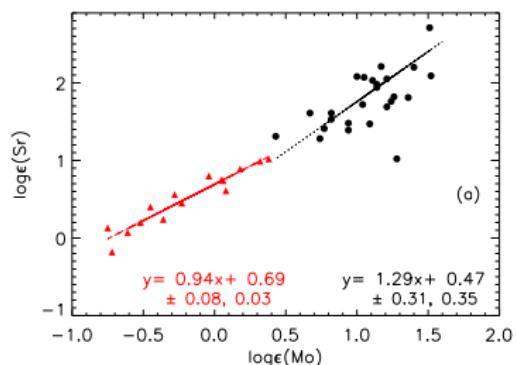


Large uncertainties and scatter found between Sr-Mo and Ag-Mo.
 Can this be improved by fitting two processes/contributions?



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Mo – weak s or LEPP? → Not LEPP



Hansen et al., 2014

Telescopes

Abundances

Applications

Heavy elements

2. r-process

Yields

Meteorites

Separating processes

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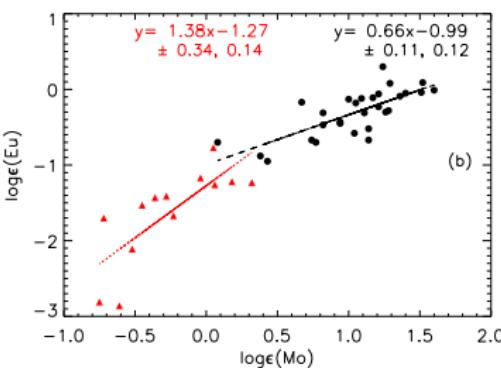
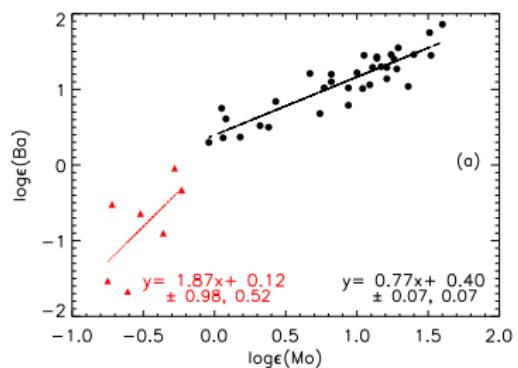
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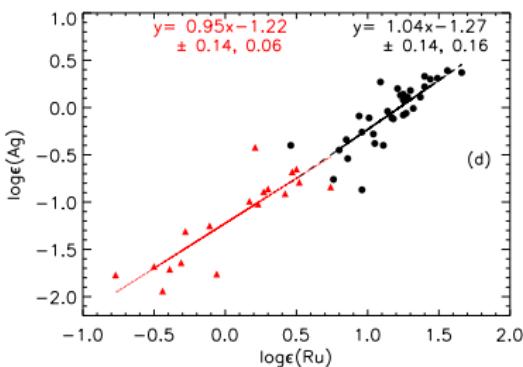
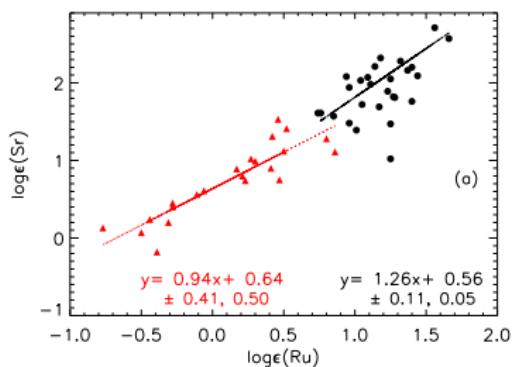
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Sr - Eu

Mo – main s or r?



Rμ - weak s or LEPP? → LEPP!



Hansen et al., 2014

Telescopes

Abundances

Applications

Heavy elements

2. r-process

Yields

Meteorites

Separating processes

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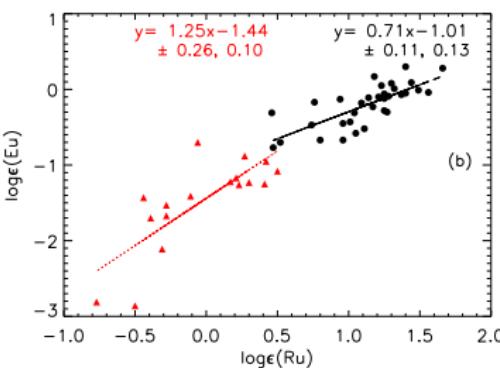
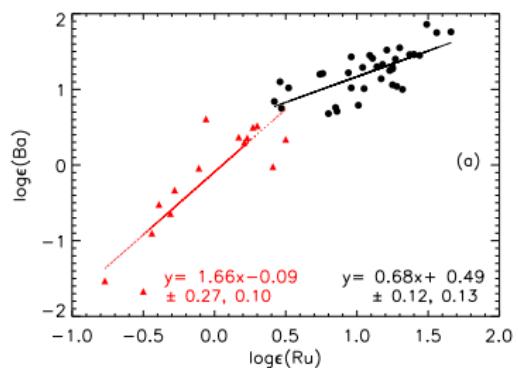
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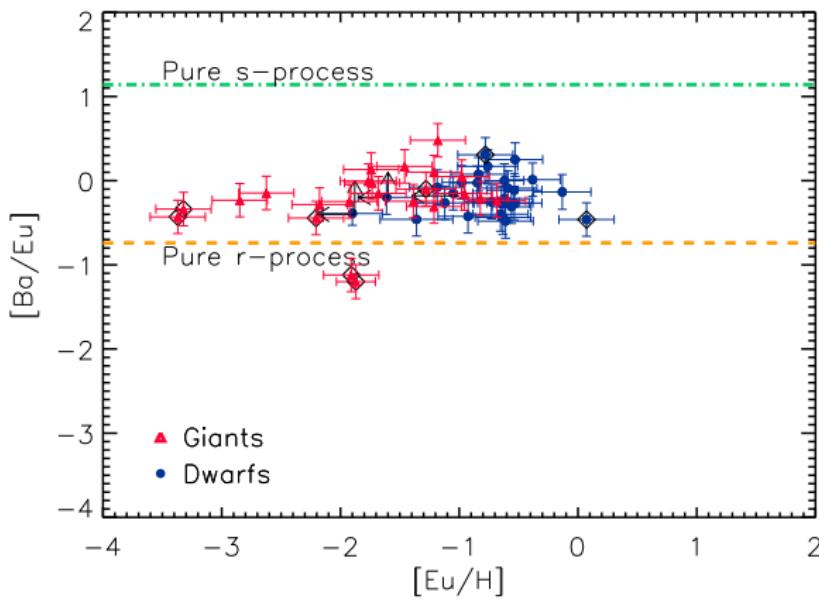
Sr - Eu

Ru – main s or r?



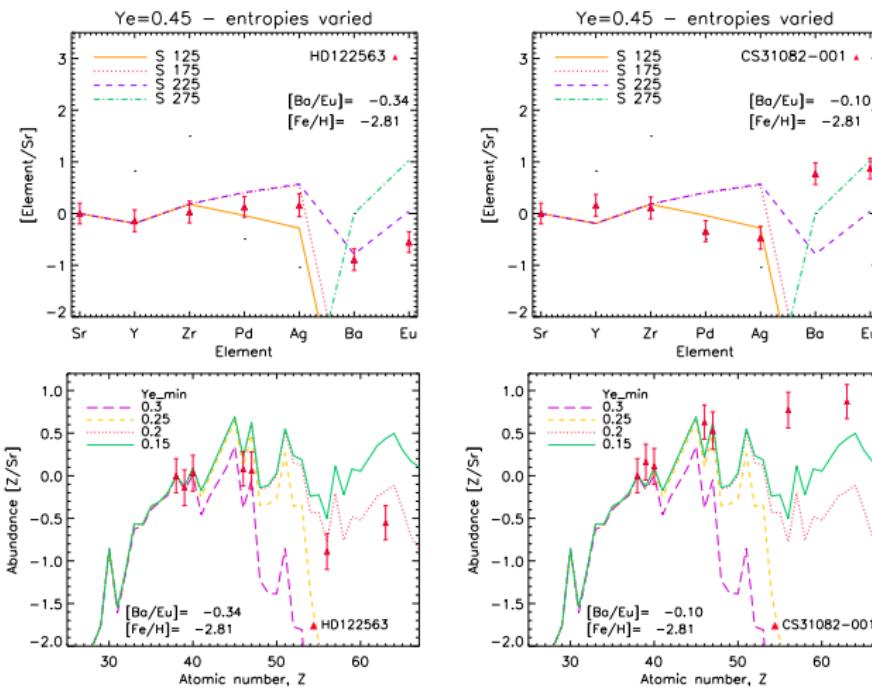
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Pure r-process yields (Hansen et al., 2012)



r-poor vs r-rich stars: HD122563 & CS31082-001

(Honda et al., 2006, Hill et al., 2002 & Hansen et al., 2012)



Summary: Observational indications of a 2nd r-process

- Ag, Pd, and Ru correlate - they are produced by the same process (LEPP/weak r/...)
- Ru+Ag do not correlate with weak s-process elements; Sr & Y
- Ru+Ag do not correlate with Ba (main s-process at solar metallicity) or Eu (94% main r-process element; Arlandini et al 1999)
- Mo is less weak r/LEPP and more weak+main s (some main r and p-process)
- → Mo is a very mixed element; it has more in common with the lighter than the heavy elements.



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Presolar grains

Isotopic abundances needed → presolar grains from meteorites?



Periodic Table of the Elements

© www.elementsdatabase.com

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ga	Pt	No	Pm	Sr	Eu	Ge	Tb	Dy	Ho	Er	Tm	Yb	Lu
90	91	92	93	94	95	96	97	98	99	100	101	102	103
Tb	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

Blue: ground based observations, green: space, yellow: isotopic abundances

Presolar grains: r-, s-, and p-process contributions to Mo and Ru

(Dauphas +2004)

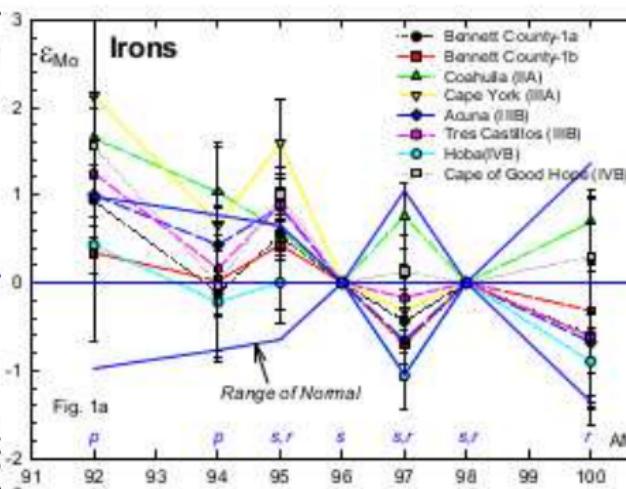
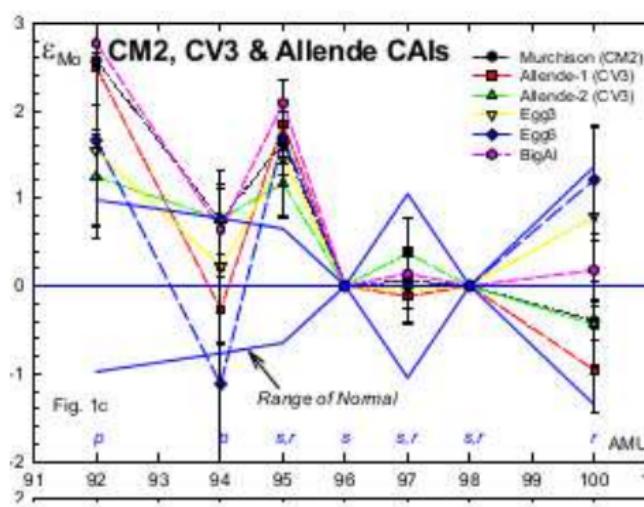
Element/Isotope	92	94	95	96	97	98	100
Mo							
Ru	96	98	99	100	101	102	104
Process	p	p	s + r	s	s + r	s + r	r

Presolar grains can be enriched by only one AGB star.

Anomalies in abundances can therefore indicate a heterogeneous gas which in turn means that the nebula/cloud was not uniformly mixed – or general variations of ^{x}Mo due to variations in the contribution from process x to the gas....

Presolar grains

Anomalies - improved method!



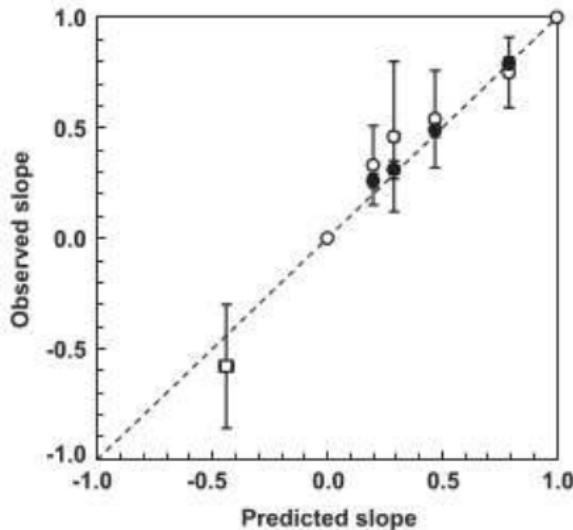
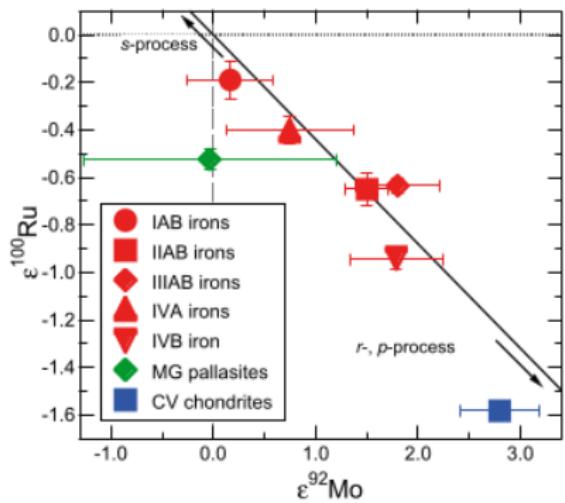
Chen et al, 2004



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Presolar grains

The slope of these correlations match s-process predicted slopes
 (for bulk meteorites). Dauphas et al, 2004



Dauphas et al therefore believe that the reason for anomalies is variations in the s-process (but cannot fully exclude r- and p-process decoupling).



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Earth

- The Mo-Ru (cosmic) correlation reflects a mixing line between pure s and Solar composition. All meteorites follow this correlation.
- The Earth also follows this cosmic correlation - this is quite interesting because:
 - Ru is highly siderophile and therefore sinks into the core
 - Mo is moderately siderophile and will stay in the mantle (like noble metals) → The same Mo-Ru correlation for meteorites would not a priori be expected for the Earth's mantle....
 - Since the Mo-Ru correlation is true for the Earth's mantle, Ru must be delivered to the mantle after the core formed by a late accretion event which was of similar composition to the gas that first enriched the mantle in Mo.



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Earth

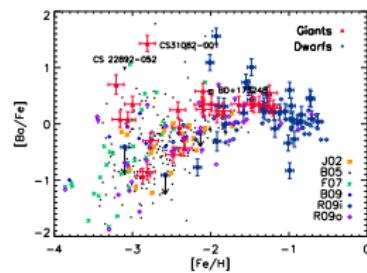
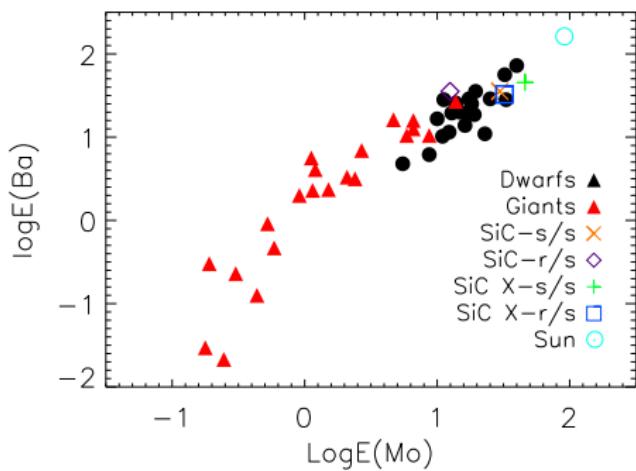
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Presolar grains

s-process in grains and stars



Solid symbols are stars, open symbols SiC grains Hansen et al., 2014, Pellin et al., 2006, Nicolussi et al., 1997

Conclusion

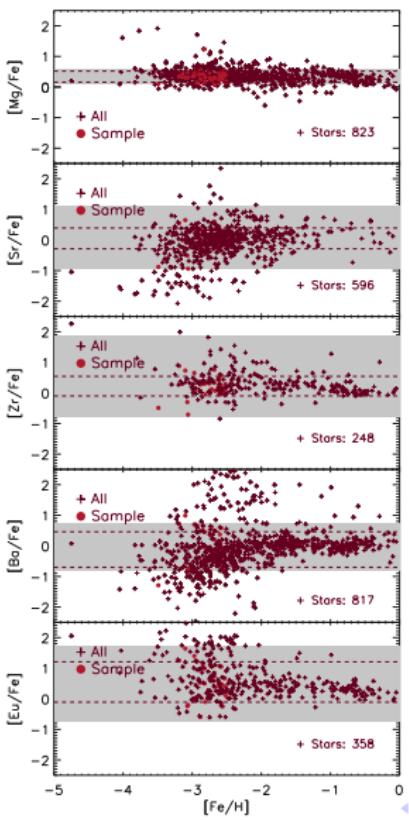
- A second process is needed to explain Ag, Ru & Pd
- This second “LEPP” is different from the s-processes and the main r-process
- Mo is produced by all processes - p,s, and r - this is detectable
- Mo and Ru are important heavy elements as they can trace various formation processes and thereby provide information on the formation of stars, meteorites, and Earth.
- Two processes seem sufficient to explain the stellar abundances and their scatter within the uncertainty (0.32dex) - may be too large = could hide other contributions
- Room for improvement:
 - 3D self-consistent SN models,
 - optimized yield predictions,
 - 3D+NLTE abundance corrections for heavy elements and
 - mixing processes in the ISM.

Material for discussion: Observational indicators for formation processes -

- 1) Correlations
- 2) star-to-star abundance scatter
- 3) Abundance pattern from observations
- 4) Uncertainties
- 5) CEMP stars



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Telescopes

Abundances

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2. r-process

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Meteorites

Separating processes

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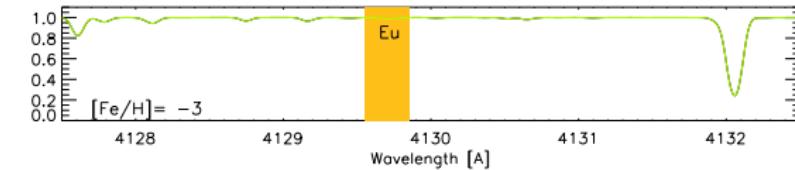
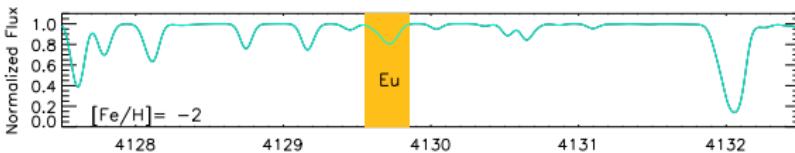
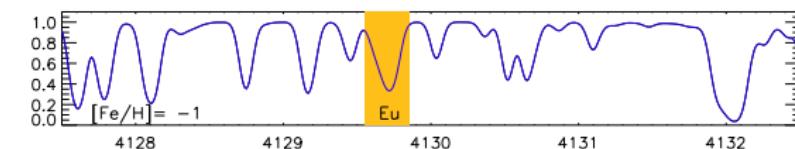
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r/L EPP



Observational abundance biases (Hansen et al., 2014 subm. to ApJ)



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From this sample we eliminate stars with:

- $[\text{Fe}/\text{H}] < -2.5$ - removes most s-process contamination
 - $[\text{C}/\text{Fe}] < 0.9$ - removes most CEMP stars
 - $[\text{Ba}/\text{Fe}] < 1.0$ - removes CEMP-s and Ba-rich binaries
 - Min. 5 abundance detections (i.e., not upper limits)
 - $[\text{C}/\text{N}] < -0.4$ and $[\text{N}/\text{Fe}] > 0.5$ - removes self-enriched stars

Assumptions:

There are 3 robust processes:

r-process, LEPP, P-component.

M1 :

r=CS22892-052, LEPP=HD122563

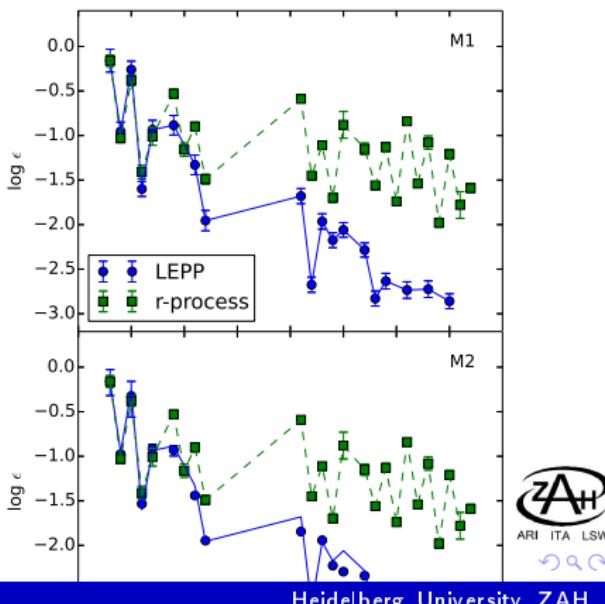
M2: r=CS22892-052.

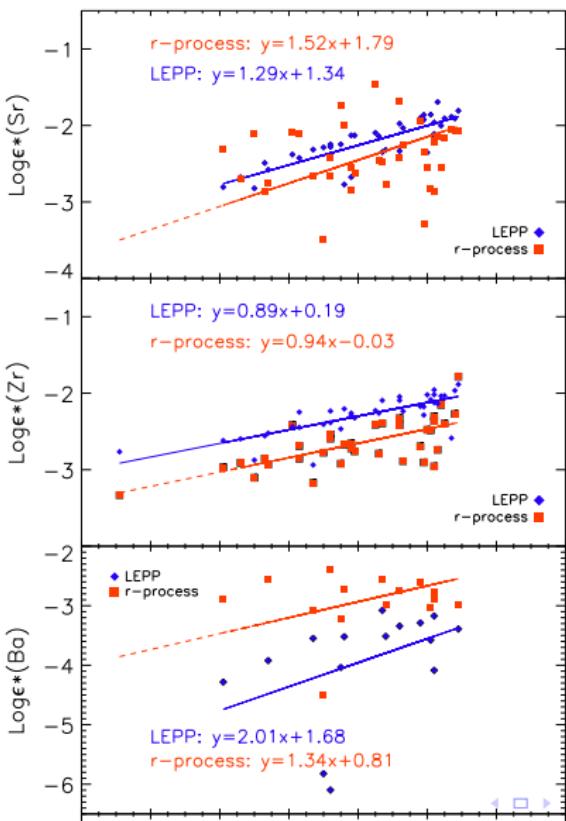
r+LEPP = HD122563

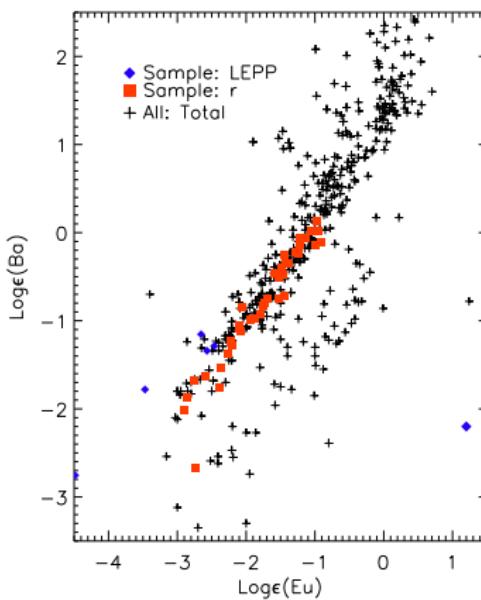
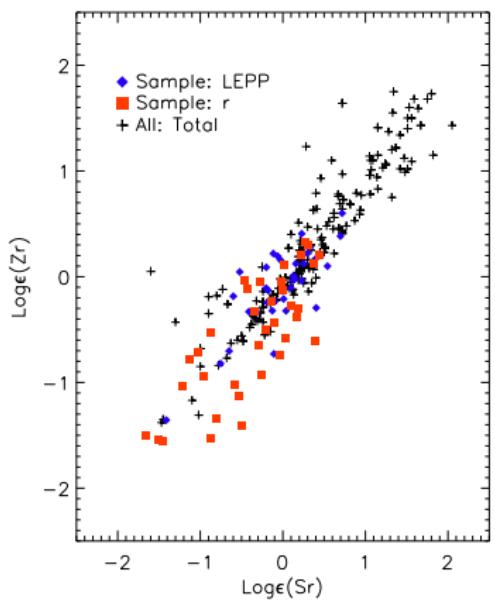
M3: r+LEPP=CS22892-052,

r+LEPP=HD122563

- all stars are mixed



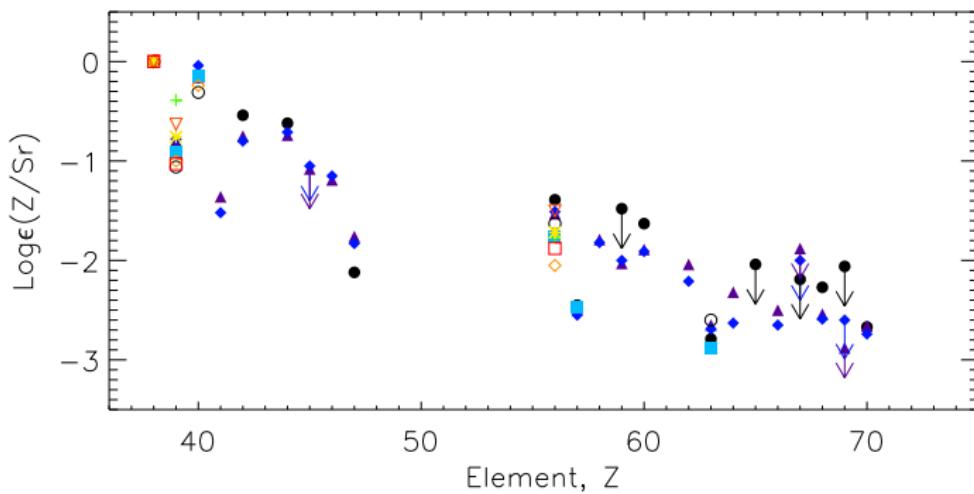




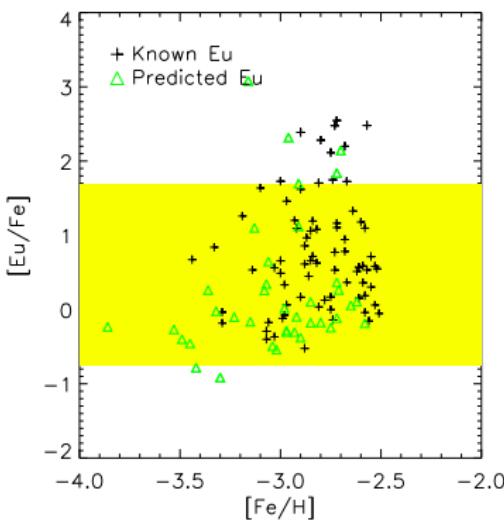
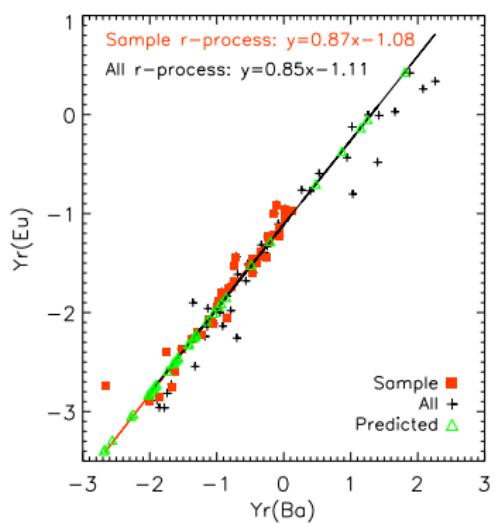
Robustness of the processes! (Hansen et al., 2014 subm. to ApJ)



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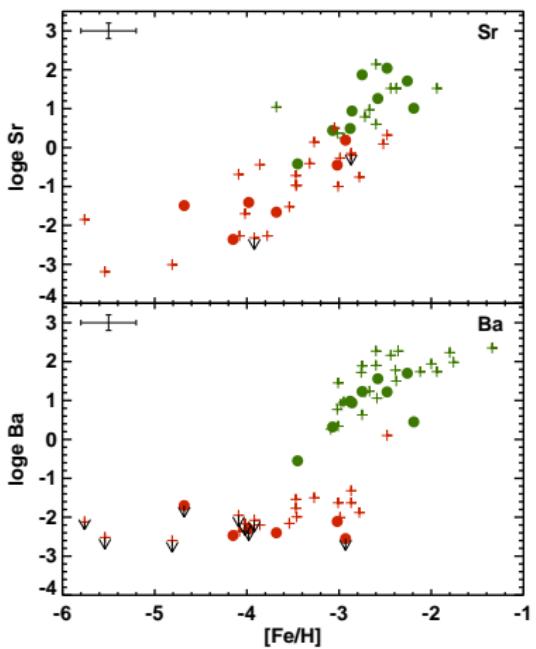
Robustness of the LEPP!



Robustness of the r-process!



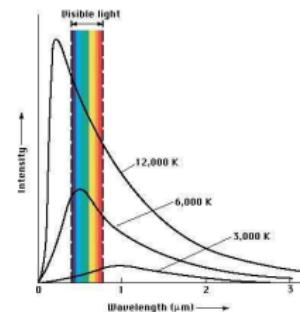
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CEMP stars

Two ways of deriving abundances:

- Equivalent width and synthetic spectra
- We need to know the stellar parameters:
Temperature, gravity,
metallicity and velocity (small scale)
- Model atmosphere (e.g. MARCS)
and synthetic spectrum code (e.g. MOOG)
- Assumptions: 1D, LTE –
one local temperature, black body radiation
(Planck), Maxwellian velocity distribution,
Boltzmann and Saha describe excitation and ionisation
- Line lists with atomic and molecular
information
(excitation potential and log gf)



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Temperature, gravity and metallicity

- The color of a star depends on two factors: Temperature and metallicity

- Color (V-K) calibration:

$$T = a + b(V - K) + c(V - K)^2 + d(V - K)[Fe/H] + \dots$$

- Excitation potential - based on Fe lines (NLTE sensitive)

- Parallax/distance (π):

$$\log \frac{g}{g_{Sun}} = \log \frac{M}{M_{Sun}} + 4 \frac{T}{T_{Sun}} + 0.4V_o + 2\log(\pi) + corrections$$

- Ionisation equilibrium from Fe lines (NLTE sensitive)

- Metallicity ([Fe/H]) from equivalent widths of Fe lines